11.1 The distribution of particle speeds of a certain hypothetical gas is given by

$$N(v)dv = Ave^{-v/v_0}dv,$$

where A and v_0 are constants.

- (a) Determine A so that $f(v) \equiv N(v)/N$ is a true probability density function; i.e., $\int_0^\infty f(v) dv = 1$. Sketch f(v) versus v.
- (b) Find \bar{v} and $v_{\rm rms}$ in terms of v_0 .
- (c) Differentiate f(v) with respect to v and set the result equal to zero to find the most probable speed v_m .
- (d) The standard deviation of the speeds from the mean is defined as

$$\sigma \equiv \left[\overline{(v - \bar{v})^2} \right]^{1/2},$$

where the bar denotes the mean value. Show that

$$\sigma = \sqrt{\bar{v}^2 - (\bar{v})^2}$$

in general. What is σ for this problem?

- 11.3 What is the number density of molecules at a temperature of 77 K in an ultra-high vacuum at 10^{-10} torr?
- 11.6 At what values of the speed does the Maxwell speed distribution have half its maximum value? Give your answer as a constant time $(k_B T/m)^{1/2}$.